

1	Group 1		Group 2											Group 18							
1	1 <b>H</b> Hydrogen 1.007 94																			2 <b>He</b> Helium 4.002 602	
2	3 <b>Li</b> Lithium 6.941	4 <b>Be</b> Beryllium 9.012 182																			
3	11 <b>Na</b> Sodium 22.989 770	12 <b>Mg</b> Magnesium 24.3050																			
4	19 <b>K</b> Potassium 39.0983	20 <b>Ca</b> Calcium 40.078	21 <b>Sc</b> Scandium 44.955 910	22 <b>Ti</b> Titanium 47.867	23 <b>V</b> Vanadium 50.9415	24 <b>Cr</b> Chromium 51.9961	25 <b>Mn</b> Manganese 54.938 049	26 <b>Fe</b> Iron 55.845	27 <b>Co</b> Cobalt 58.933 200												
5	37 <b>Rb</b> Rubidium 85.4678	38 <b>Sr</b> Strontium 87.62	39 <b>Y</b> Yttrium 88.905 85	40 <b>Zr</b> Zirconium 91.224	41 <b>Nb</b> Niobium 92.906 38	42 <b>Mo</b> Molybdenum 95.94	43 <b>Tc</b> Technetium (98)	44 <b>Ru</b> Ruthenium 101.07	45 <b>Rh</b> Rhodium 102.905 50												
6	55 <b>Cs</b> Cesium 132.905 43	56 <b>Ba</b> Barium 137.327	57 <b>La</b> Lanthanum 138.9055	72 <b>Hf</b> Hafnium 178.49	73 <b>Ta</b> Tantalum 180.9479	74 <b>W</b> Tungsten 183.84	75 <b>Re</b> Rhenium 186.207	76 <b>Os</b> Osmium 190.23	77 <b>Ir</b> Iridium 192.217												
7	87 <b>Fr</b> Francium (223)	88 <b>Ra</b> Radium (226)	89 <b>Ac</b> Actinium (227)	104 <b>Rf</b> Rutherfordium (261)	105 <b>Db</b> Dubnium (262)	106 <b>Sg</b> Seaborgium (266)	107 <b>Bh</b> Bohrium (264)	108 <b>Hs</b> Hassium (277)	109 <b>Mt</b> Meitnerium (268)												

**Key:**

Atomic number — 6

Symbol — **C**

Name — Carbon

Average atomic mass — 12.0107

- Hydrogen
- Semiconductors (also known as metalloids)

- Metals**
- Alkali metals
  - Alkaline-earth metals
  - Transition metals
  - Other metals

- Nonmetals**
- Halogens
  - Noble gases
  - Other nonmetals

			Group 13	Group 14	Group 15	Group 16	Group 17	Group 18			
			5 <b>B</b> Boron 10.811	6 <b>C</b> Carbon 12.0107	7 <b>N</b> Nitrogen 14.0067	8 <b>O</b> Oxygen 15.9994	9 <b>F</b> Fluorine 18.998 4032	10 <b>Ne</b> Neon 20.1797			
			13 <b>Al</b> Aluminum 26.981 538	14 <b>Si</b> Silicon 28.0855	15 <b>P</b> Phosphorus 30.973 761	16 <b>S</b> Sulfur 32.065	17 <b>Cl</b> Chlorine 35.453	18 <b>Ar</b> Argon 39.948			
Group 10	Group 11	Group 12	28 <b>Ni</b> Nickel 58.6934	29 <b>Cu</b> Copper 63.546	30 <b>Zn</b> Zinc 65.409	31 <b>Ga</b> Gallium 69.723	32 <b>Ge</b> Germanium 72.64	33 <b>As</b> Arsenic 74.921 60	34 <b>Se</b> Selenium 78.96	35 <b>Br</b> Bromine 79.904	36 <b>Kr</b> Krypton 83.798
			46 <b>Pd</b> Palladium 106.42	47 <b>Ag</b> Silver 107.8682	48 <b>Cd</b> Cadmium 112.411	49 <b>In</b> Indium 114.818	50 <b>Sn</b> Tin 118.710	51 <b>Sb</b> Antimony 121.760	52 <b>Te</b> Tellurium 127.60	53 <b>I</b> Iodine 126.904 47	54 <b>Xe</b> Xenon 131.293
			78 <b>Pt</b> Platinum 195.078	79 <b>Au</b> Gold 196.966 55	80 <b>Hg</b> Mercury 200.59	81 <b>Tl</b> Thallium 204.3833	82 <b>Pb</b> Lead 207.2	83 <b>Bi</b> Bismuth 208.980 38	84 <b>Po</b> Polonium (209)	85 <b>At</b> Astatine (210)	86 <b>Rn</b> Radon (222)
			110 <b>Ds</b> Darmstadtium (281)	111 <b>Uuu*</b> Unununium (272)	112 <b>Uub*</b> Ununbium (285)	113 <b>Uut*</b> Ununtrium (284)	114 <b>Uuq*</b> Ununquadium (289)	115 <b>Uup*</b> Ununpentium (288)			

A team at Lawrence Berkeley National Laboratories reported the discovery of elements 116 and 118 in June 1999. The same team retracted the discovery in July 2001. The discovery of elements 113, 114, and 115 has been reported but not confirmed.

63 <b>Eu</b> Europium 151.964	64 <b>Gd</b> Gadolinium 157.25	65 <b>Tb</b> Terbium 158.925 34	66 <b>Dy</b> Dysprosium 162.500	67 <b>Ho</b> Holmium 164.930 32	68 <b>Er</b> Erbium 167.259	69 <b>Tm</b> Thulium 168.934 21	70 <b>Yb</b> Ytterbium 173.04	71 <b>Lu</b> Lutetium 174.967
95 <b>Am</b> Americium (243)	96 <b>Cm</b> Curium (247)	97 <b>Bk</b> Berkelium (247)	98 <b>Cf</b> Californium (251)	99 <b>Es</b> Einsteinium (252)	100 <b>Fm</b> Fermium (257)	101 <b>Md</b> Mendelevium (258)	102 <b>No</b> Nobelium (259)	103 <b>Lr</b> Lawrencium (262)

The atomic masses listed in this table reflect the precision of current measurements. (Values listed in parentheses are the mass numbers of those radioactive elements' most stable or most common isotopes.)

\* The systematic names and symbols for elements greater than 110 will be used until the approval of trivial names by IUPAC.