

Nuevas técnicas para el desarrollo de materiales mediante el uso de antimateria

Positron Annihilation Spectroscopy

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²²Na source

²²Na is produced by bombardment of Mg target with suitable projectiles such as protons, deuterons or α particles.



• 1.27 MeV γ appears almost simultaneously with positron - can be used as start event for lifetime spectroscopy

Positron inside matter



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Wave-particle duality

After thermalization:

- Classical point of view: following the diffusion theory, the positron diffuses tens to hundreds of nanometers (diffusion length) and then annihilates with an electron of the crystal.
- Quantum mechanics point of view: the positron wave function is expanded inside a defect free crystal before annihilation with an electron wave.
- Possible agreement: the positron diffusion length is of the order of the "radius" of the expansion volume of the wave function.



Crystal lattice After thermalization the positron wave function is spread over the crystal. Then, the positron annihilates with an electron. τ_b~100..200 ps







iii. Positron lifetime spectroscopy

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Positron Annihilation Lifetime Spectroscopy - PALS



 positron lifetime is measured in a ²²Na laboratory setup as the time difference between the 1.27 MeV "start" and the annihilation 0.511 MeV "stop" gamma quanta.

Positron Annihilation Lifetime Spectroscopy - PALS



- Positron lifetime is measured as time difference between 1.27 MeV quantum (β^+ decay) and 0.511 MeV quanta (annihilation process)
- PM...photomultiplier; SCA...single channel analyzer (constant-fraction type); TAC...time to amplitude converter; MCA... multi channel analyzer

Time resolution - PALS

Data analysis

$$S(t) = \left(\sum_{i=1}^{n} \frac{I_i}{\tau_i} e^{-\frac{t}{\tau_i}}\right) \star R(t) + B$$
$$\sum_{i=1}^{n} I_i = 1$$
$$\frac{1}{\tau} = \lambda = \pi r_e^2 c \int |\psi_+(\mathbf{r})|^2 n_-(\mathbf{r}) d\mathbf{r}$$

 λ : positron annihilation rate $\psi_{+}(\mathbf{r})$ is the positron wave function. $n_{-}(\mathbf{r})$ is the electron density.

1 J. Kansy, Microcomputer program foranalysis of positron annihilation lifetimespectra. Nucl. Instrum. Methods Phys. A **374**, 235 (1996).

The analysis of the spectrum is performed thanks to specific software suites, such as LT software¹.

It is possible to resolve the deconvolution of the resolution function and analyze the components.



Typical positron lifetime spectrum in an epoxy-based industrial adhesive. The positron implantation energy was 18 keV according to a mean implantation depth of 4 μm . The spectrum can be decomposed into three exponentials with three different lifetimes.



LiCoO₂ Samples with Carbon Coating



(a) SEM micrograph of the PVDF/LiCoO₂ grains.

(b) Schematic of graphite/LiCoO₂ heterostructure.

(c) Positron lifetime spectra of $LiCoO_2$ cathodes with different grain boundaries, with (blue) and without (orange) graphite. The slopes are the inverse of the positron lifetime of each component.



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First PALS results



	Cathode	Thickness (µm)	τ ₁ (ps)	τ ₂ (ps)	$I_1(\%)$	I ₂ (%)	τ (ps)
Exp.	LiCoO2 + PVDF + Graphite	114 (2)	152 (2)	319 (2)	74 (2)	26 (2)	197 (3)
Exp.	LiCoO2 + PVDF	115 (2)	145 (2)	1	100	/	145 (2)
Cal.	LiCoO2 (Perfect Crystal)	/	131	/	/	1	/

Thickness, positron lifetime components, relative intensities and average positron lifetime obtained in the LiCoO₂ cathode with and without graphite. Calculated data is based on GGA + LDA.



G. Pagot, V. Di Noto, K. Vezzù, B. Barbiellini, V. Toso, A. Caruso, M. Zheng, X. Li, R. Ferragut. *Quantum view of Li-ion high mobility at carbon-coated cathode interfaces*. iScience 26, 105794 (2023).

PALS & BES correlation







Real permittivity (ε' , A) and conductivity (σ' , B) spectra on frequency of the four different samples containing LiCoO₂, PVDF and: without carbon (*No C*, blue circles), carbon Super P (*SP*, green triangles), carbon nanospheres (*XC*, red diamonds), and carbon nanotubes (*NT*, black squares). Markers represent experimental data and dashed lines are the fitting results. Concentration of effective mobile lithium ions (n_{Li^+eff} , C) and lithium diffusion coefficient (D_{Li}^+ , D) as a function of the carbon employed into the cathode electrode. Error bars (2 *sigma*) are calculated on the basis of the fitting error and experimental data accuracy. Dotted lines in C represent the total lithium ion concentration in the electrodes with (red line) and without (blue line) carbon.

Battery test



CR2032 battery tests results. Discharge curves are obtained at different current rates (0.5C, 1C, 2C and 4C) with configuration Li | LiPF6 1 M in EC:DMC 1:1 | *cathode*, where *cathode* corresponds to each sample employing SP (green), XC (red) and NT (black) carbon materials (a).

Durability tests in the first 150 cycles at a constant current rate (0.5C) for each sample employing SP (green), XC (red) and NT (black) carbon materials (b).

First-principles calculations





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(b) Sandwich model SW-G \oplus LCO (d) e⁺ inject into the SW supercell VC/

VC/SW-G⊕LCO heterostructures

Band structures



(a) Electronic band structure and density of states (DOS) of LiCoO₂ bulk. Total electronic state, Li, Co, O are shown in black dashed line, lime (b) Electronic band structure of free standing graphene layers.





Fig. 2 Electronic band structure and density of states (DOS) of SW-G \oplus LCO at the equilibrium distance $d_1 = 2.4$ Å. Total electronic state, C, Li, Co, O are shown in black dashed line, gray, lime green, blue and orange, respectively. High symmetry line are calculated along Γ , M, K, Γ . Dirac point of graphene is located at the K-point and shifts downward by 1.0 eV with respect to the Fermi level shown in red dashed line.

Schematic Diagram of Schottky Barrier



(a) Schematic diagram of Schottky Barrier



(b) Different interaction types and fermi level position within the band gap for different types of semiconductors.

Energy (eV)





Calculated plane-averaged charge density difference of SW-G⊕LCO perpendicular to (001) plane at the equilibrium distance. The corresponding isosurface of charge density difference is the inset. Orange (green) region represents charge accumulation (depletion).

Results of the positron calculations: (a) graphene stack (GS) bulk, (b) six-layer GS slab and (c) SW-G⊕LCO heterostructure. The maximum positron probability density occurs at 1.8 Å from the graphene surface. Ground-state positron densities are presented along c direction. (d) Threedimensional plot of calculated WDA positron total potential in a GS slab.



iv. Doppler broadening of the annihilation radiation

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Birth γ -ray 1.27 MeV Δt 1. Positron lifetime Sample 10⁵ 511keV peak 10⁴ Compton-edge positron source Counts 22-Na diffusion (100 nm) 1022keV pile-up valley \ 10² 100 µm 10¹ 0 200 400 600 800 1000 Thermalization E (keV) ~1-10 ps 3. Doppler broadening Energy spectrum (GeHP) 0.511 MeV ± ΔE , $\Delta E = p_z c/2 = p_L c/2$

 p_L follows the longitudinal direction along the sample and the detector axis

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Insulator

Metal

Semiconductor

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Momentum distribution of the positron-electron pair before annihilation ρ (**p**)

$$\rho(\mathbf{p}) = \pi r_e^2 c \sum_i \left| \int d\mathbf{r} e^{-i\mathbf{p}\cdot\mathbf{r}} \psi_+(\mathbf{r}) \psi_i(\mathbf{r}) \right|^2$$

 $\psi_{+}(\mathbf{r})$ is the positron wave function.

 $\psi_i(\mathbf{r})$ is the electron wave function.



Experimental setup of Doppler-broadening spectroscopy.









$$S = A_{centr} / A_{tot}$$

|E - 511 keV| < 0.85 keV







Figure 3. Evolution of the *S* parameter as a function of the positron implantation energy (mean implantation depth, upper frame) in $LiCoO_2$ (green symbols), in $Li_{0.5}CoO_2$ (brown symbols) and in a graphite reference sample (black dashed line). The VEPFIT fitting (green and brown dashed lines) gives an estimation of the positron diffusion length.

- J. Nokelainen, B. Barbiellini, J. Kuriplach, S. Eijt, R. Ferragut, X. Li, V. Kothalawala, K. Suzuki, H. Sakurai, H. Hafiz, K. Pussi, F. Keshavarz, A. Bansil. Identifying Redox Orbitals and Defects in Lithium-Ion Cathodes with Compton Scattering and Positron Annihilation Spectroscopies: A Review. Condens. Matter 7 47 (2022).
- G. Pagot, V. Toso, B. Barbiellini, R. Ferragut, V. Di Noto. Positron Annihilation Spectroscopy as a Diagnostic Tool for the Study of LiCoO₂ Cathode of Lithium-Ion Batteries. Condens. Matter 6 28 (2021).

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S-parameter LiCoO₂ (LCO) cathodes. Spectra of *SP*, *XC*, *NT* and *No C* cathodes.

No C: without carbon. Only the PVDF polymer binder SP: Super P carbon XC: carbon nano-spheres NT: carbon nano-tubes

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Defects in GaN films grow by LEPEVPE





 $\Delta E = p_L c/2$ (energy shift)

 p_L follows the longitudinal direction along the sample and the detector axes



K.G. Lynn et al., Phys. Rev. Lett. 38 (1977) 241; P. Asoka-Kumar et al., Phys. Rev. Lett. 77 (1996) 2097

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Coincidence Doppler Broadening of the annihilation radiation

 $E_{\gamma 1} + E_{\gamma 2} = 2 m_0 c^2 = 1022 \text{ keV}$

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Coincidence Doppler Broadening of the annihilation radiation

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FIG. 3. The comparison of the experimental and theoretical annihilation probability densities for Al, Si, and Ge. The area under the experimental data is normalized to unity. The calculated curves have been normalized to the quantity λ_c/λ_{tot} , where λ_c and λ_{tot} are the annihilation rate with core electrons and the total annihilation rate, respectively.

2 September 1996



p_L follows the longitudinal direction along the sample and the detector axes

FIG. 4(color). The spectrum for different elements after normalizing to Al: (a) experiment and (b) theory. The theoretical curves for $p_L < 20 \times 10^{-3} m_0 c$ [dashed line in (b)] are not accurate.



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CDB results



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Momentum distribution:

$$N(p_L) = -\kappa \, p_L \frac{dI(p_L)}{dp_L}$$

CDB results



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Momentum distribution:

$$N(p_L) = -\kappa \, p_L \frac{dI(p_L)}{dp_L}$$

Linear combination:

 $N(p_L) = (\beta - 1) \cdot N_{LCO_{Noc}}(p_L) + \beta \cdot N_C(p_L)$

The average lifetime $\bar{\tau}_E$ can be estimated by

$$\lambda_E = \bar{\tau}_E^{-1} \cong (1 - \beta) \cdot \tau_1^{-1} + \beta \cdot \tau_2^{-1}$$

Results of the linear combination proposed to test the consistence of the lifetime and CDB measurements. $\bar{\tau}_m$ is the measured value.

Cathode	β	$ar{ au}_E(ps)$	$ar{ au}_m(ps)$
SP (Super P)	0.45 (3)	201 (4)	197 (3)
XC (nanospheres)	0.55 (4)	232 (5)	231 (4)
NT (nanotubes)	0.60 (3)	235 (4)	233 (3)

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CDB simulations





Computed Doppler spectra for (a) LiCoO₂. (b), (c) and (d) show the graphite, nanosphere (C60), nanotube (CNT-12), respectively. Valence orbitals are shown in solid lines, and contributions from core electrons are indicated with dashed lines.

Positron localization



(a) The positron density of CNT-12 is mapped, with areas of high density depicted in purplish-red and the electron density of the carbon nanotubes shown in yellow. (b) Within the white dashed square in (a), the positron potential energy is shown in green, and the positron wavefunctions between the nanotubes are marked on isosurfaces. (c) A three-dimensional mapping of the positron density on the 001 plane.

CDB simulations

Momentum distributions



Linear Combination of DFT and EXP

 a^* β^* Adjusted

 SP
 Graphite
 0.496 0.504 99.731%

 XC
 C60
 0.426 0.574 99.656%

 NT
 CNT-80.344 0.656 99.966%

 CNT-12
 0.250 0.750 99.983%

Momentum distributions, N (P_L) for calculates LCO and carbon, along with their linear combination, compared with experimental results from diverse samples. The momentum profiles for different measured sample types in (a) SP, (b) XC, and (c-d) NT, are represented by black symbols. The linear combination of the calculated spectra is denoted by the red solid line.

Positrons as a quantum probe

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Quantum view of Li-ion high mobility at carbon coated cathode interfaces

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iScience 26, 105794 (2023).

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